

FIG. 1

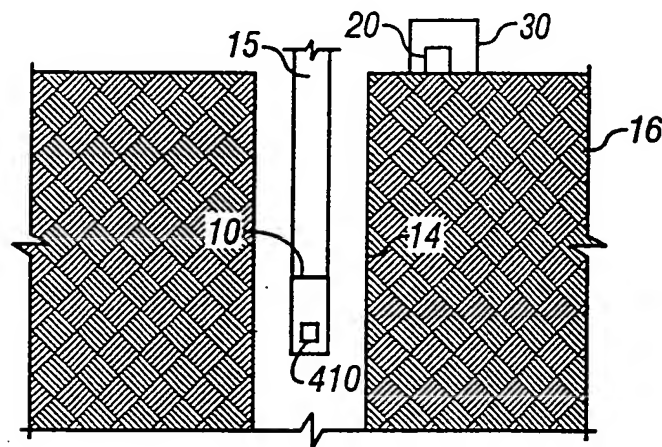


FIG. 2

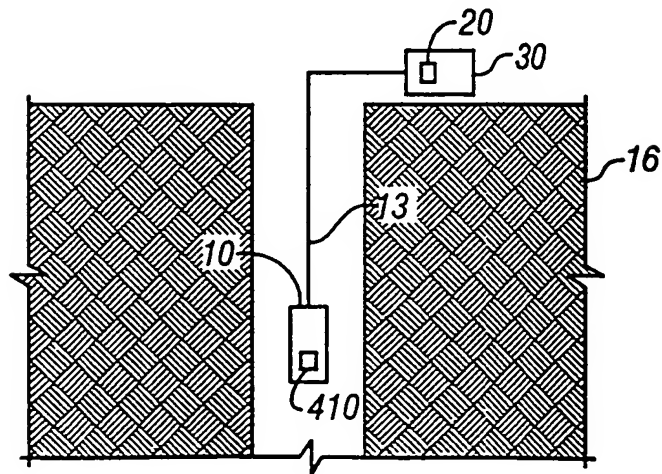


FIG. 3

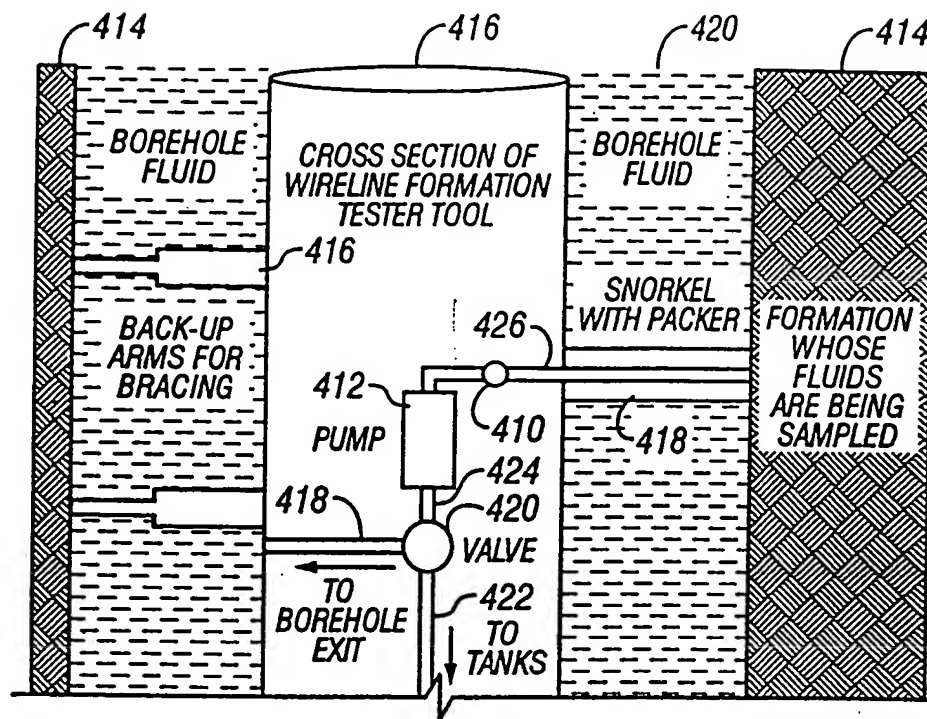


FIG. 4

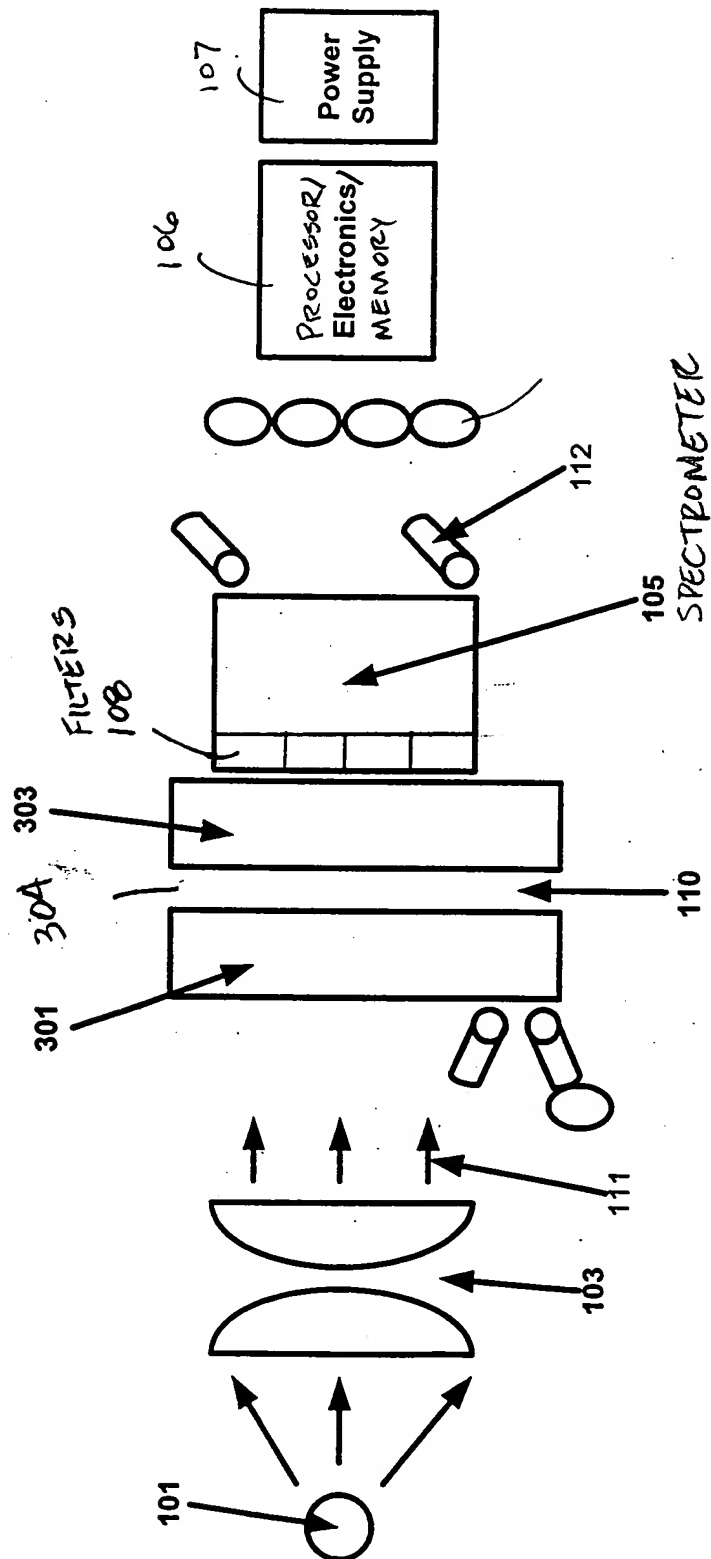


FIG. 5

Equations Correlating Weight Fraction Methane in Mixtures of Crude Oil and Methane to Optical Absorbance and Temperature

FIG. 6

Methane Weight Fraction = METHWTF = $B0 + B1 \cdot \text{Var1} + B2 \cdot \text{Var2} + B3 \cdot \text{Var3} + B4 \cdot \text{Var4} \dots$

Regression Summary for Dependent Variable: METHWTF
R = .98093203 R² = .96222765 Adjusted R² = .96151158
F(4,211) = 1343.8 p < 0.0000 Std.Error of estimate: .04992

	B
Var1 = SQ70_82	0.06514 = B0 = Intercept
Var2 = TEMP_C	11.1756 = B1
Var3 = SRSA1682	0.00087 = B2
Var4 = SRSA1670	-2.66167 = B3
	2.63245 = B4

Regression Summary for Dependent Variable: METHWTF
R = .98190316 R² = .96413381 Adjusted R² = .96327986
F(5,210) = 1129.0 p < 0.0000 Std.Error of estimate: .04876

	B
Var1 = SRSA1670	0.03143 = B0 = Intercept
Var2 = SRSA1682	2.53111 = B1
Var3 = SQ70_82	-2.55766 = B2
Var4 = TEMP_C	11.9135 = B3
Var5 = TEMP_SQR	0.0019 = B4
	-6.2E-06 = B5

SQ70-82 = SQUARE(Absorbance_at_1670_nm - Absorbance_at_1682_nm)
SRSA1670 = SQRT(Absorbance_at_1670_nm)
SRSA1682 = SQRT(Absorbance_at_1682_nm)
TEMP_C = Temperature in Degrees Centigrade
TEMP_SQR = Square of Temperature in Degrees C

Equation for Density of Methane [g/cc] as
a Function of Pressure and Temperature
from 100 - 30,000 psia and 75 - 200 C
is fitted by Adj. R² = .99911359

Equation for Optical Absorbance per mm of
Methane as Function of Density and Wavelength
at 11 nm FWHM, Center λ range of 1668-1684 nm,
for 100-30,000 psia, and 75 - 200 C,
is fitted by Adj. R² = .94145159

	B
P	2.771E-03 = Intercept
P ²	2.480E-05
P ³	-1.120E-09 for Pressure in psi
T ²	1.808E-14
(P/T)	-1.308E-07 for Temperature in C
(P/T) ²	1.455E-03
(P/T) ³	-4.922E-06
	5.934E-09

where, WaveNumber = 10 000 000 / λ [nm]

Equations Relating Gas Oil Ratio, GOR, to Weight Fraction of Methane, f_M, and Stock Tank Density, ρ_O, of Oil

1 bbl = 0.159 m³ = 5.615 cu ft = 42 U.S. gal
1 Standard Cubic Foot (SCF) of Methane Gas at 14.7 psia & 60°F is 0.042358 lbs = 19.21327 grams.
Density of Methane at 60 F and 14.7 psia is 0.0006787 gr/cc = 0.042358 lbm/ft³.
Letting V = Volume, W = Weight, ρ = Density, and using subscripts M for Methane and O for Oil,
GOR = $V_{\text{Methane}} [\text{SCF}] / V_{\text{Oil}} [\text{bbls}] = \{W_M / (19.21 \text{ g/SCF})\} / \{(W_O / \rho_O) (1 \text{ bbl} / 158.983 \text{ cc})\}$
Letting f_M = Weight Fraction of Methane,
GOR = $8274.62 \rho_O / (1 / f_M - 1)$

f_M = $W_M / (W_M + W_O) = \rho_M V_M / (\rho_M V_M + \rho_O V_O)$ so $W_O = W_M / (1 / f_M - 1)$ which substitutes into above.
f_M = $1 / (1 + 8274.62 \cdot \rho_O / \text{GOR})$ where W_G and W_O are in grams, ρ_O is in g/cc, and f_M = Wt. Frac. of Methane

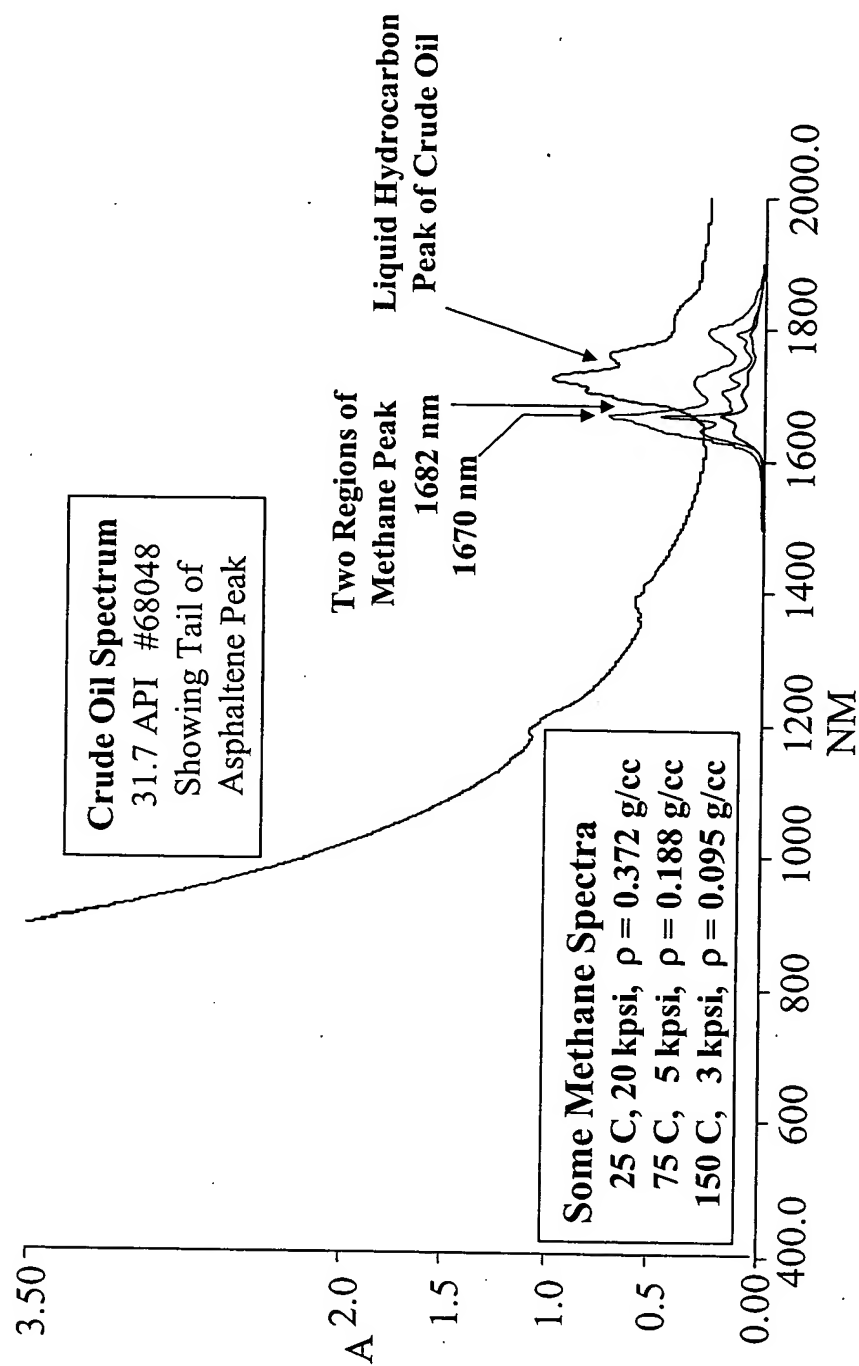


Figure 7

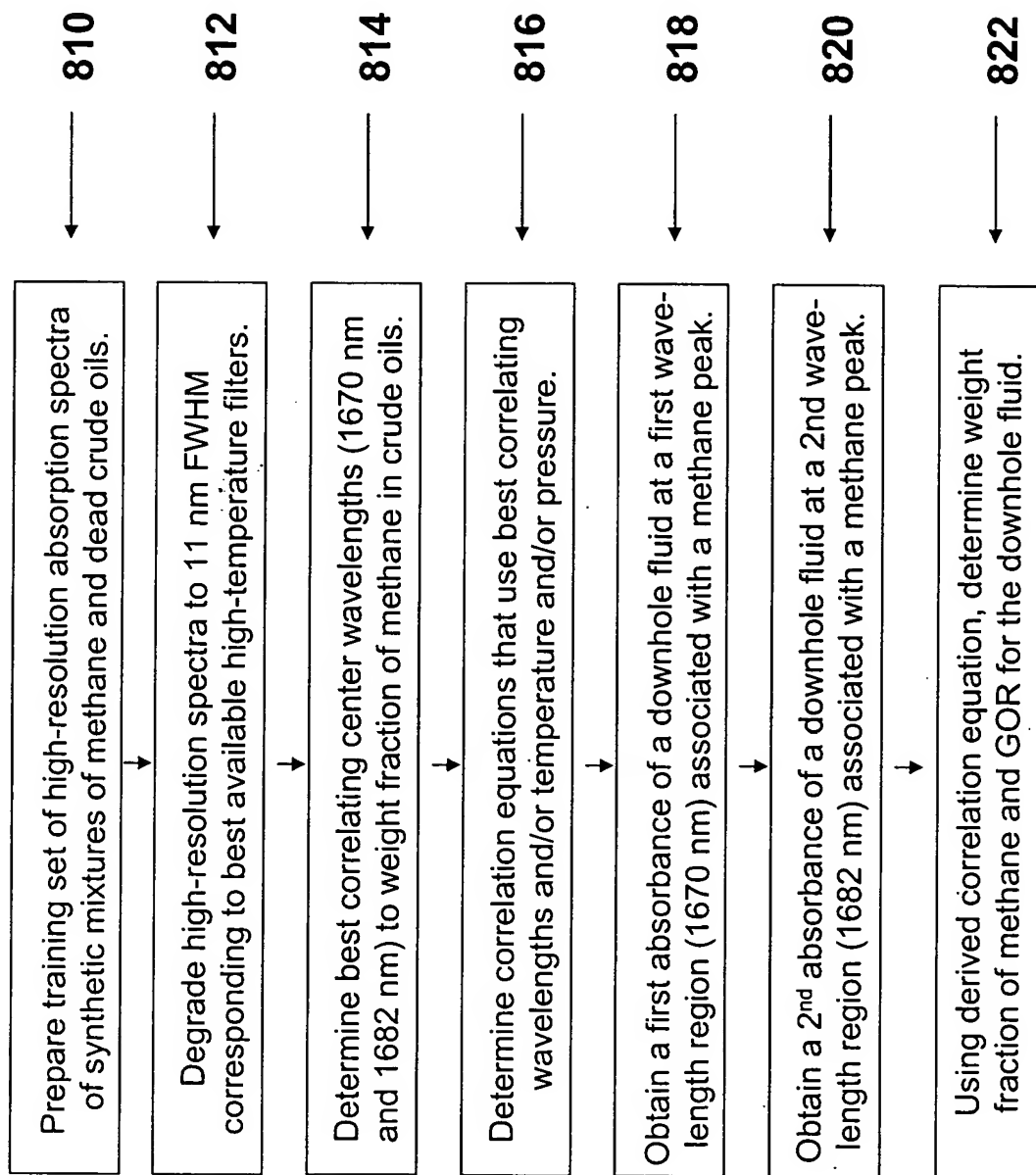


Figure 8